

## Molecular Geometries and Bond Angles

Chemical Formula	AXE Formula	Pair Arrangement	Molecular Geometry	Bond Angle
BeCl <sub>2</sub>	AX <sub>2</sub> E <sub>0</sub>	Linear	Linear	180°
BCl <sub>3</sub>	AX <sub>3</sub> E <sub>0</sub>	Trigonal Planar	Trigonal Planar	120°
CH <sub>4</sub>	AX <sub>4</sub> E <sub>0</sub>	Tetrahedral	Tetrahedral	109.5°
NH <sub>3</sub>	AX <sub>3</sub> E <sub>1</sub>	Tetrahedral	Trigonal Pyramidal	107°
H <sub>2</sub> O	AX <sub>2</sub> E <sub>2</sub>	Tetrahedral	Bent	104.5°
PCl <sub>5</sub>	AX <sub>5</sub> E <sub>0</sub>	Trigonal Bipyramidal	Trigonal Bipyramidal	120° and 90°
SeF <sub>4</sub>	AX <sub>4</sub> E <sub>1</sub>	Trigonal Bipyramidal	See-Saw	120° and 90°
BrF <sub>3</sub>	AX <sub>3</sub> E <sub>2</sub>	Trigonal Bipyramidal	Tee	120° and 90°
TeF <sub>6</sub>	AX <sub>6</sub> E <sub>0</sub>	Octahedral	Octahedral	90°
XeF <sub>4</sub>	AX <sub>4</sub> E <sub>2</sub>	Octahedral	Square Planar	90°

## Molecular Geometries and Dipole Moments

Chemical Formula	AXE Formula	Molecular Geometry	Dipole Moment
BeCl <sub>2</sub>	AX <sub>2</sub> E <sub>0</sub>	Linear	No
BCl <sub>3</sub>	AX <sub>3</sub> E <sub>0</sub>	Trigonal Planar	No
CH <sub>4</sub>	AX <sub>4</sub> E <sub>0</sub>	Tetrahedral	No
NH <sub>3</sub>	AX <sub>3</sub> E <sub>1</sub>	Trigonal Pyramidal	Yes
H <sub>2</sub> O	AX <sub>2</sub> E <sub>2</sub>	Bent	Yes
PCl <sub>5</sub>	AX <sub>5</sub> E <sub>0</sub>	Trigonal Bipyramidal	No
SeF <sub>4</sub>	AX <sub>4</sub> E <sub>1</sub>	See-Saw	Yes
BrF <sub>3</sub>	AX <sub>3</sub> E <sub>2</sub>	Tee	Yes
TeF <sub>6</sub>	AX <sub>6</sub> E <sub>0</sub>	Octahedral	No

## Electron Pair Arrangements and Bond Hybridization

Chemical Formula	AXE Formula	Pair Arrangement	Bond Hybridization
BeCl <sub>2</sub> and C <sub>2</sub> H <sub>2</sub>	AX <sub>2</sub> E <sub>0</sub>	Linear	sp
BCl <sub>3</sub> and C <sub>2</sub> H <sub>4</sub>	AX <sub>3</sub> E <sub>0</sub>	Trigonal Planar	sp <sup>2</sup>
CH <sub>4</sub> and C <sub>2</sub> H <sub>6</sub>	AX <sub>4</sub> E <sub>0</sub>	Tetrahedral	sp <sup>3</sup>
NH <sub>3</sub>	AX <sub>3</sub> E <sub>1</sub>	Tetrahedral	sp <sup>3</sup>
H <sub>2</sub> O	AX <sub>2</sub> E <sub>2</sub>	Tetrahedral	sp <sup>3</sup>
PCl <sub>5</sub>	AX <sub>5</sub> E <sub>0</sub>	Trigonal Bipyramidal	sp <sup>3</sup> d
SeF <sub>4</sub>	AX <sub>4</sub> E <sub>1</sub>	Trigonal Bipyramidal	sp <sup>3</sup> d
BrF <sub>3</sub>	AX <sub>3</sub> E <sub>2</sub>	Trigonal Bipyramidal	sp <sup>3</sup> d
TeF <sub>6</sub>	AX <sub>6</sub> E <sub>0</sub>	Octahedral	sp <sup>3</sup> d <sup>2</sup>